

ATTY. DKT. 28950.0002

STATES PATENT AND TRADEMARK OFFICE

Serial No. 10/530,889

In re the Application of

Confirmation No. 3235

M. YAMAGUCHI et al

Group Art Unit 1753

Filed: April 11, 2005

For:

POSITIVE ELECTRODE ACTIVE MATERIAL FOR BATTERY, METHOD

FOR PREPARING ELECTROLYTIC MANGANESE DIOXIDE, AND BATTERY

REQUEST FOR ISSUANCE OF CORRECTED FILING RECEIPT

Commissioner for Patents U.S. Patent & Trademark Office **Customer Window** Randolph Building 501 Dulany Street Alexandria, VA 22314

Sir:

The official Filing Receipt contains an error in the spelling of the third inventor's given name through mistake by the Office. Thus, it is respectfully requested that a corrected Filing Receipt be issued to include the correct spelling of the third inventor's name as shown in red on the attached copy of the Filing Receipt, as follows:

Change "Oichi Numata" to --Koichi Numata--.

Respectfully submitted,

STEPTOE & JOHNSON LLP

Roger W. Parkhurst

Reg. No. 25,177

1330 Connecticut Ave., NW Washington, D.C. 20036

Tel: 202-429-3000



27890



UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS PC. Dox 1450 Alexandra, Vignia 22313-1450

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FILING OR 371 APPL NO **ART UNIT** FIL FEE REC'D ATTY.DOCKET NO **DRAWINGS** TOT CLMS IND CLMS (c) DATE 10/530,889 04/11/2005 1753 1030 16 1 28950.0002

CONFIRMATION NO. 3235

FILING RECEIPT

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Date Mailed: 07/06/2005

1330 CONNECTICUT AVENUE, N.W. WASHINGTON, DC 20036

STEPTOE & JOHNSON LLP

Receipt is acknowledged of this regular Patent Application. It will be considered in its order and you will be notified as to the results of the examination. Be sure to provide the U.S. APPLICATION NUMBER, FILING DATE, NAME OF APPLICANT, and TITLE OF INVENTION when inquiring about this application. Fees transmitted by check or draft are subject to collection. Please verify the accuracy of the data presented on this receipt. If an error is noted on this Filing Receipt, please mail to the Commissioner for Patents P.O. Box 1450 Alexandria Va 22313-1450. Please provide a copy of this Filing Receipt with the changes noted thereon. If you received a "Notice to File Missing Parts" for this application, please submit any corrections to this Filing Receipt with your reply to the Notice. When the USPTO processes the reply to the Notice, the USPTO will generate another Filing Receipt incorporating the requested corrections (if appropriate).

Applicant(s)

Munetoshi Yamaguchi, Hiroshima, JAPAN; Yasuhiro Ochi, Hiroshima, JAPAN; Oichi Rumata, Hiroshima, JAPAN; Shigeo Hirayama, Hiroshima, JAPAN;

Koichi

Power of Attorney: The patent practitioners associated with Customer Number 27890.

Domestic Priority data as claimed by applicant

This application is a 371 of PCT/JP03/02640 03/06/2003

Foreign Applications

JAPAN 2002-299301 10/11/2002

Projected Publication Date: 10/06/2005

Non-Publication Request: No

Early Publication Request: No

Title

Positive plate active material forcell, method for producing electrolytic manganese dioxide, and



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REQUEST FOR CORRECTED PUBLICATION UNDER 37 CFR 1.221(b)

Commissioner for Patents
U.S. Patent & Trademark Office
Customer Window
Randolph Building
501 Dulany Street
Alexandria, VA 22314

Sir:

The spelling of the third inventor's given name in Patent Publication No. US-2005-0233215-A1 is in error through the mistake of the Office, which error is apparent from the Office records. The spelling of the third inventor's name should be <u>KOICHI NUMATA</u>.

In view of the foregoing, it is respectfully requested that this request for a corrected or revised patent application publication be granted.

Respectfully submitted,

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(19) United States

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(54) POSITIVE PLATE ACTIVE MATERIAL FORCELL, METHOD FOR PRODUCING ELECTROLYTIC MANGANESE DIOXIDE, AND CELL

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				C25	B 1/21
(52)	U.S. Cl.	429/22	4: 423	/605: 2	05/539

(57) ABSTRACT

A positive electrode active material for battery consisting of electrolytic manganese dioxide and a battery wherein this positive electrode active material is used, this electrolytic manganese dioxide being such that, when it is heated and the weight loss thereof is measured, the weight loss at 200° C. to 400° C. is not less than 2.7 wt %. The amount of weight lost in this temperature range is proportional to the quantity of bound waters inside the electrolytic manganese dioxide.

